



# Optimization of the β-Aminoester Class of Factor Xa Inhibitors. Part 1: P<sub>4</sub> and Side-Chain Modifications for Improved In Vitro Potency

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Abstract—A systematic modification of the  $C_3$  side-chain of the  $\beta$ -aminoester class of factor Xa inhibitors and a survey of  $P_4$  variations is described. These changes have resulted in the identification of sub-nanomolar inhibitors with improved selectivity versus related proteases. Coagulation parameters (i.e., APTT doubling concentrations) are also improved. © 2002 Elsevier Science Ltd. All rights reserved.

The inhibition of the serine protease component of the prothrombinase complex, factor Xa (fXa), is generally recognized as a viable strategy for the development of novel antithrombotic agents. The unique position of fXa at the convergence of the intrinsic and extrinsic pathways, its singular role in thrombin activation and in the amplification of the coagulation cascade combine to make it an attractive intervention point. In response, a growing number of publications have appeared describing small molecule fXa inhibitors and several comprehensive reviews have been published.

In a previous publication,<sup>4</sup> we reported the discovery and initial SAR of a novel series of  $\beta$ -aminoester derived fXa inhibitors 1 and 2. Therein, we established the requirement for the ester moiety, the optimal stereochemistry at the  $C_2$  and  $C_3$  positions (R,R) and in a preliminary way, examined the SAR of the biaryl  $P_4$  moiety.

A modeling study of 2 in the active site of des1-45 factor Xa predicted that the benzamidine moiety binds in the  $S_1$  pocket while the biphenyl group fills the aromatic pocket  $(S_4)$ . The model also suggested that the  $C_1$ 

methoxycarbonyl group is positioned near the catalytic triad, however, our biochemical studies did not support this view.<sup>4</sup> Neither were we able to elucidate the exact role and importance of the C<sub>3</sub> styryl side-chain. Subsequent X-ray analysis indicated that the methoxycarbonyl group occupies a shallow groove remote from the catalytic triad.<sup>5,6</sup>

In this communication, we describe the systematic variation of the  $C_3$   $\beta$ -aminoester side-chain using a common synthetic approach. In addition, the SAR of the  $P_4$  group was further elucidated. This work resulted in the identification of nanomolar inhibitors of factor Xa with selectivity against related trypsin-like serine proteases. The subsequent article expands this theme to identify the  $\beta$ -aminoester **FXV673** as a development candidate for acute thrombosis.  $^6$ 

## Chemistry

The  $\beta$ -aminoesters **4** bearing the requisite *R*-configuration at  $C_3$  are prepared by Arndt–Eistert homologation of the corresponding Boc-D-amino acid derivatives

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and proceed via the intermediary diazomethyl ketones  $3.^8$  As expected, the homologated esters 4 are obtained with retention of configuration at the asymmetric center (Scheme 1). Subsequent deprotonation of 4 and alkylation with 3-cyanobenzyl bromide affords the adducts 5 as single diastereomers (>95% de as determined by  $^1$ H NMR). The Boc protecting group is then removed under standard conditions providing the alkylated  $\beta$ -aminoesters 6, which are then coupled to appropriate 4-biphenylcarboxylic acids. Finally, conversion of the nitrile moiety of intermediates 7 to their corresponding benzamidines is carried out using Pinner/ammonolysis chemistry delivering the target  $\beta$ -aminoester inhibitors 8, 9a, and 9c–g in good yield.

The suitably protected aminomethylbiphenyl carboxylic acid 18 required for compounds 9b and 13–16 is prepared via palladium catalyzed cross coupling of 3-bromobenzonitrile and ethyl-4-iodobenzoate as outlined in Scheme 2. Subsequent reduction affords 17 whose amine function is Boc protected. Base-catalyzed hydrolysis of the ester provides the requisite carboxylic acid 18, for coupling to amines 6 (Scheme 1). The nitrile moiety of coupled product 11 is converted in a three-step process, to amidines 12; final Boc deprotection yields the desired inhibitors 9b and 13–16.

The racemic inhibitor 10 was prepared as outlined in Scheme 3. Acylation of commercially available  $\beta$ -alanine methyl ester hydrochloride with 4-biphenylcarbonyl chloride followed by alkylation with 3-cyanobenzyl bromide afforded intermediate nitrile; which is transformed to amidine 10 using standard conditions.

## Discussion

Compound 1 is representative of the  $\beta$ -aminoester class of fXa inhibitors originally discovered in our laboratories. Further optimization of this series began with a systematic examination of the  $C_3$  substituent R (Table 1). Initial efforts were focused on compounds bearing a biphenyl moiety in the putative  $P_4$  position. Compounds were assayed against human factor Xa, thrombin and trypsin as previously described. In comparing inhibitors 1, 2, 8, 9a, and 10, it is evident that large hydrophobic substituents are not particularly favored at  $C_3$ . Complete removal (R = H) resulted in only a modest reduction in fXa potency for the racemate 10.

Concurrent work had established the desirability of appending substituents to the distal ring of the biphenyl group;<sup>5</sup> Table 2 summarizes the data. The marked

Scheme 1. Reagents and conditions: (a) i-BuCOCl, THF,  $-30\,^{\circ}$ C; (b) CH<sub>2</sub>N<sub>2</sub>, ether,  $0\,^{\circ}$ C; (c) PhCO<sub>2</sub>Ag, NEt<sub>3</sub>, MeOH; (d) (i) LHMDS, THF,  $-78\,^{\circ}$ C to  $-25\,^{\circ}$ C; (ii) 3-cyanobenzyl bromide, THF,  $-78\,^{\circ}$ C to rt; (e) TFA, CH<sub>2</sub>Cl<sub>2</sub>,  $0\,^{\circ}$ C; (f) Ar–Ph-4–CO<sub>2</sub>H, TBTU, iPr<sub>2</sub>NEt, DMF, rt; (g) (i) H<sub>2</sub>S, pyridine, NEt<sub>3</sub>; (ii) MeI, acetone; (iii) NH<sub>4</sub>OAc; (h) HCl, MeOH, rt; (ii) NH<sub>3</sub>, MeOH, reflux; (i) 18, TBTU, iPr<sub>2</sub>NE<sub>t</sub>, DMF, rt.

**Scheme 2.** Reagents and conditions: (a) *n*BuLi, THF, -78 °C; ZnCl<sub>2</sub>; 3-bromo-benzonitrile, Pd(Ph<sub>3</sub>P)<sub>2</sub>Cl<sub>2</sub>; (b) H<sub>2</sub>, Rh on alumina, NH<sub>3</sub>, MeOH; (c) Boc<sub>2</sub>O, DMAP, DMF, rt; (d) aqueous NaOH.

$$\underset{O}{\text{MeO}} \underset{NH_2}{\overset{\text{A, b,}}{\longrightarrow}} \underset{H_2N}{\overset{\text{MeO}}{\longrightarrow}} \underset{10}{\overset{\text{O}}{\longrightarrow}} \underset{N}{\overset{\text{MeO}}{\longrightarrow}} \underset{N}{\overset{\text{O}}{\longrightarrow}} \underset{N}{\overset{\text{O}}{\longrightarrow}} \underset{N}{\overset{\text{MeO}}{\longrightarrow}} \underset{N}{\overset{\text{O}}{\longrightarrow}} \underset{N}{\overset{\text{O}}{\longrightarrow}} \underset{N}{\overset{\text{MeO}}{\longrightarrow}} \underset{N}{\overset{\text{O}}{\longrightarrow}} \underset{N}{\overset{N}} \underset{N}{\overset{N}}{\overset{N}} \underset{N}{\overset{N}} \underset{N}{\overset{N}}{\overset{N}} \underset{N}{\overset{N}} \underset{N}} \underset{N}{\overset{N}} \underset{N}} \underset{N}{\overset{N}} \underset{N}{\overset{$$

**Scheme 3.** Reagents and conditions: (a) 4-biphenylcarbonyl chloride, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, rt; (b) LHMDS, THF, -78 °C then 3-cyanobenzyl bromide, -78 °C to rt; (c) (i) HCl, MeOH, rt; (ii) NH<sub>3</sub>, MeOH, reflux.

increase in potency observed for 9b was attributed to the charged nature of the benzyl amine, since a number of dibasic inhibitors make use of the  $S_4$  cation hole of factor Xa for productive interaction. The Although this interpretation was supported by entries 9c and 9d, the result for 9c clearly indicates that a positive charge is not a prerequisite for good activity and H-bonding interactions of the carboxamide with the  $S_4$  pocket may be involved. It appears that a number of different factors are important for binding and a variety of strategies can be employed to optimize interaction with the  $S_4$  pocket.

Variations at  $C_3$ , which employ the aminomethylbiphenyl  $P_4$  group, allowed us to explore the SAR around this position in more detail (13–16 and 9b). The

Table 1. SAR of C<sub>3</sub> side-chain

Compd	R	R′	$K_i^{10}$ (nM)			
			fXa	fIIa	Tryp.	
1 <sup>a</sup>	·	}—н	21	810	67	
<b>2</b> <sup>a</sup>	·	}—н	9.4	954	221	
8	,	{—н	11	709	75	
9a	₹—Me	}—н	5.3	3250	69	
10 <sup>b</sup>	<b></b> ₹—н	}—н	20	> 3950	230	
13	, ·	`NH₂	4.0	650	36	
14	\	∑, NH <sub>2</sub>	0.40	990	38	
15	<i>`</i> <	5NH <sub>2</sub>	8.0	> 3950	300	
16	<b>\</b>	NH <sub>2</sub>	0.70	2100	63	
9b	ζ—Me	`NH₂	0.9	~2900	69	

<sup>&</sup>lt;sup>a</sup>Synthesis described previously.<sup>4</sup>

isobutyl substituent (14) was the most effective while steric crowding closer to the  $C_3$  branching point (15) resulted in decreased fXa inhibitory activity. Although the  $C_3$  substituent is required for stereocontrol at  $C_2$  during synthesis of  $\beta$ -aminoesters (Scheme 1), the potency advantage for larger alkyls over methyl is marginal. Given the simplicity and ready availability of the  $\beta$ -aminobutyric acid precursor 4 (i.e., R = Me), compounds with a methyl group at the  $C_3$  position (e.g., 9a-g) became the focus of further efforts.

The relative activity of several inhibitors verses related trysin-like serine proteases is shown in Table 3. Although very good selectivities have been achieved against thrombin (fIIa), APC and tPA, selectivity against trypsin and plasmin is modest. In comparing compounds  $\bf 9a$ ,  $\bf 9b$ , and  $\bf 2$ , the  $\bf C_3$  methyl side-chain derivatives are no less selective than the styryl. The incorporation of the basic aminomethyl substitutent in  $\bf P_4$  appears to impart better potency as well as across the board selectivity (cf.,  $\bf 9a$  and  $\bf 9b$ ).

Table 2. SAR of biaryl P4 ligands

Compd	Ar	$K_i^{10}$ (nM)			
		fXa	fIIa	Tryp.	
9a	7	5.3	~3250	69	
9b	NH <sub>2</sub>	0.9	2920	69	
9c	r <sup>rt</sup> NMe <sub>2</sub>	1.0	2660	95	
9d	NHCO <sub>2</sub> Et	> 1200	> 4000	> 2900	
9e	O NH <sub>2</sub>	0.5	> 4000	90	
9f	NH <sub>2</sub>	2.0	> 4000	42	
9g	ОН	69	> 4000	ND	

**Table 3.** In vitro<sup>10</sup> comparisons of selected  $\beta$ -aminoester fXa inhibitors

Compd	fXa $K_i$ (nM)		Ratio: $K_i$ enz/ $K_i$ fXa			2 X APTT (μM)		
		fIIa	Tryp.	APC	Plasmin	tPA	Human plasma	Rat plasma
2	9.4	100	24	> 2050	80	> 965	ND	ND
9a 9b	5.3 0.9	610 3250	13 77	> 3500 650	87 230	> 1600 > 9650	3.9 0.33	5.05 1.7

<sup>&</sup>lt;sup>b</sup>Racemic.

Anticoagulant activity in plasma from human and rat was measured for select compounds. Inhibitors **9a** and **9b** were effective anticoagulants in vitro and reflected the trend established against free factor Xa (Table 3). Subtle species differences were observed perhaps due to structural variations in enzyme.

## Conclusion

A systematic study on the effects of the  $C_3$  branching substituent on the potency and selectivity of the  $\beta$ -aminoester series of factor Xa inhibitors has been carried out. The results reveal that small alkyl substituents are preferred at this position and the  $C_3$  substituent could be completely removed with only a modest reduction in fXa inhibitory potency. Significant increases in potency could also be achieved by substitutions on the distal  $P_4$  phenyl ring.

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- 8. Alternatively, derivative **4** (R = Me) may be readily prepared on a large scale via conjugate addition of Li (R)-N-( $\alpha$ -methylbenzyl)benzylamide to methyl crotonate followed by debenzylation using Pearlman's catalyst. See: Davies, S. G.; Ichihara, O. *Tetrahedron: Asymm.* **1991**, 2, 183. Final Boc protection (BOC<sub>2</sub>O, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>) then provides  $\beta$ -aminoester 4 (R = Me) in good yield.
- 9. Satisfactory spectral data were obtained for all compounds. Final products were purified by C18 reverse-phase HPLC, eluting with a 0.1% TFA  $H_2O/CH_3CN$  gradient, lyophilized and isolated as powders.
- 10. In vitro enzyme assays and activated partial thromboplastin times (APTT) were performed as described in: Choi-Sledeski, Y. M.; McGarry, D. G.; Green, D. M.; Mason, H. J.; Becker, M. R.; Davis, R. S.; Ewing, W. R.; Dankulich, W. P.; Manetta, V. E.; Morris, R. L.; Spada, A. P.; Cheney, D. L.; Brown, K. D.; Colussi, D. J.; Chu, V.; Heran, C. L.; Morgan, S. R.; Bentley, R. G.; Leadley, R. J.; Maignan, S.; Guilloteau, J. P.; Dunwiddie, C. T.; Pauls, H. W. J. Med. Chem. 1999, 42, 3572.